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Sparknotes: sat chemistry: intra- and intermolecular forces

The two main types of intermolecular forces that exist between molecules are dipole-dipole forces (including hydrogen bonds) and London dispersion forces.

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Long range intermolecular interactions between the

Abstract. Long range interactions between the ground state alkali diatomics , , , and NaK-NaK are examined. Interaction energies are first determined from ab initio

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as implemented in the CADPAC 6.5 program package was used to calculate intermolecular interaction
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